7. Asymmetric *Diels-Alder* Cycloadditions with Chiral Carbamoyl Dienophiles

by Albert Defoin, Agnès Brouillard-Poichet, and Jacques Streith*

Ecole Nationale Supérieure de Chimie de Mulhouse, Université de Haute-Alsace, 3, rue Alfred Werner, F–68093 Mulhouse Cedex

(30.X.91)

Chiral acylnitroso dienophiles 14, which were obtained from L-proline and from D-mandelic acid, reacted with cyclohexa-1,3-diene to give the expected diastereoisomers 15 and 16 (*Scheme 2* and *Table 1*). The *d.e.* values for these *Diels-Alder* reactions were moderate; they are related to the molecular stiffness of the dienophiles. The absolute configuration of the major cycloadducts was interpreted in terms of HOMO/LUMO interactions, the approach being 'endo' and the acylnitroso dienophiles reacting from their s-cis-conformation.

Introduction. Hetero-*Diels-Alder* cycloadditions with nitroso dienophiles have been the subject of increasing interest during the last two decades [1], especially also because of the stereospecific *cis* attachement of the potential alcohol and amino functionalities at the termini of the conjugated diene [2] [3]. This methodology was used by us for the total synthesis of aminosugar derivatives [4] and by *Kresze* and coworkers for the synthesis of aminoconduritols and of aminoinositols [5] [6].

In a few cases, *Diels-Alder* cycloadditions with chiral nitroso dienophiles led to excellent asymmetric induction, *e.g.* with α -chloronitroso derivatives of epiandrosterone [7] and of D-mannose [8]. Initially, the results obtained on addition of the latter to cyclohexa-1,3-diene were not interpreted correctly in terms of absolute configuration [8]; in the mean time, the absolute configuration of the addition product was corrected as indicated in *Formula* 1 (dextrorotatory product) [9] [6].



Acylnitroso dienophiles RCO–N=O are extremely reactive and had to be prepared *in* situ by oxidation of the corresponding hydroxamic acids in the presence of the conjugated diene partners [10]. Excellent asymmetric inductions (*d.e. ca.* 98%) were achieved in cycloadditions with N-nitrosocarbonyl derivatives of C_2 symmetrical pyrrolidines **2** [11]

[12], with some drawbacks, though: *i*) the preparation of these pyrrolidines 2 required lengthy procedures; *ii*) the chiral inductors could not be removed easily from their *Diels-Alder* cycloadducts.

We describe herein some results we obtained with acylnitroso derivatives of L-proline ((S)-configuration) and of D-mandelic acid ((R)-configuration) when reacted with cyclohexa-1,3-diene¹). These acylnitroso derivatives were generated *in situ* from the corresponding hydroxamic acids 3 and 4 and 5, respectively. During our investigations, two studies were published as preliminary communications which pertain also to asymmetric *Diels-Alder* cycloadditions with chiral acylnitroso dienophiles (mandelic-acid derivatives) [14] [15].

Chiral Hydroxamic Acids. – The preparation of the L-proline-derived N-carbohydroxamic acis **3a–d** by the classical approach, *i.e.* from the corresponding carbamoyl chloride and NH₂OH [4b] [16], was unsatisfactory. Therefore, L-prolinol (**6a**), its methyl ether **6b**, and the aniline derivative **6c** were reacted with phenyl N-hydroxycarbamate (**7**) in pyridine, to give the corresponding hydroxamic acids **3a–c**, respectively (*Scheme 1*). Compound **7** was obtained from NH₂OH and phenyl carbonate or, even better, phenyl chloroformate [17].



Hydroxamic acid 3d of methyl prolinate was synthesized in excellent yield from carbamoyl chloride 8 and *O*-benzylhydroxylamine *via* 9 which was hydrogenolysed. Compounds 9 and 3d cyclized easily under mild conditions to the corresponding hydantoines 10 and 11 (reaction time and temperature had to be controlled carefully). Hydantoine 11 was the major product when carbamoyl chloride 8 was reacted with NH_2OH ; it was also formed on hydrogenolysis of the benzyloxy derivative 10.

C-Carbohydroxamic acid 4 was obtained by reaction of the mixed anhydride of N-[(tert-butoxy)carbonyl]-L-proline (12) with NH₂OH.

Hydroxamic acid **5a** was prepared in a straightforward manner from NH_2OH and the methyl ester of D-mandelic acid (**13a**) according to [18]. The preparation of methyl ether **5b** from **13b** had already been described [4b]. It should be noticed that the physical

For a preliminary communication, see [13].

properties of acid **5a** having the (R)-configuration are quite different from those described by *Kirby* and *Nazeer* for the (S)-enantiomer [14] (see *Exper. Part*).

Asymmetric Diels-Alder Cycloadditions. – Oxidation of the chiral hydroxamic acids 3–5 in CHCl₃ or MeOH with $(Pr_4N)IO_4$ in the presence of equimolar amounts of cyclohexa-1,3-diene led instantaneously to the corresponding acylnitroso dienophiles 14 which underwent Diels-Alder cycloaddition to the two expected diastereoisomeric adducts 15 and 16 (Scheme 2). The mixtures 15a/16a, 15b/16b, 15c/16c, and 15e/16e were separated.



a) For R*, see Table 1.

All diastereoisomers **15** have the (1*R*,4*S*)-configuration in their tricyclic moiety. The following methods were used to determine the relative amounts of the diastereoisomers **15** and **16**: *i*) ¹H-NMR spectroscopy in benzene (this solvent permits excellent differentiation of the olefinic H-atoms), *ii*) ¹³C-NMR spectroscopy, *iii*) HPLC, and *iv*) prep. TLC when the R_{f} values were sufficiently different (see *Table 1*). We found that the ¹³C-NMR spectra yielded the most reliable results since two sets of a large number of peaks (pair of diastereoisomers) could be compared for which the chemical environment is very similar. Comparison of the HPLC integration aeras (see *Table 1*) of the diastereoisomers **15** and **16** must be handled with great care; precise calibration for each isomer is mandatory which requires preliminary isolation and purification. Indeed, the *response factors* of **15a** and **16a**, *e.g.*, were quite different, one being twice as large as the other one.

Asymmetric induction proved to be moderate to poor with the above described chiral acylnitroso dienophiles. The best *d.e.* values were observed with the dienophiles **14a–d** obtained from the L-proline derivatives **3a–d** (*d.e.* 68% with **14b**; *Table 1*): their molecular stiffness, due to a certain degree of inhibition of free rotation at the N–CO–N=O functionality, led to higher *d.e.* values. Intramolecular H-bonding in **14f** (D-mandelic-acid series) responsible for structural rigidness, also led to a higher *d.e.* value. When the CON=O moiety can freely rotate with respect to the remainder of the molecule, *e.g.* in the case of *C*-(nitrosocarbonyl) compounds **14e**, **g**, the *d.e.* values were poor.

Similar results were recently observed with the nitroso derivatives **14f** and **14g** in the mandelic-acid series [14]. *Procter* and coworkers described some higher *d.e.* values with the nitroso derivative **14f** of **5a** (*d.e.* 68%) when the reaction was performed at lower temperature [15]. Lowering the reaction temperature did indeed increase the *d.e.* values: At r.t., **14b** produced a *d.e.* value of 68%, whereas at -70° , *d.e.* increased to 76% (*Table 1*).

Absolute Configuration of the Cycloadducts. – The absolute configuration of the cycloadducts was determined by a series of independent syntheses. Reaction of the

| Tal | ble 1. Diels-Alder Cycloa | dditions of Acylni | troso Dienophiles 14a-g (obtai | ned from the corresp | onding hydro | xamic acids) (| und Cyclohex | a-1,3-diene | |
|---|-------------------------------|--------------------|--------------------------------|----------------------|--------------------|---------------------|----------------------|-------------|----------|
| Hydroxamic | Conditions | Adducts | R* | Overall yield | Ratio 15/16 | | | | d.e. |
| acid | | : | | [%] | ¹ H-NMR | ¹³ C-NMR | HPLC | Isolation | |
| 3a (→14a) | CHCl ₃ /MeOH, r.t. | 15a/16a | $2-(OHCH_2)C_4H_7N$ | 89 | 74:26 | 76:24 | 76:24 ^ª) | 76:24 | 52 |
| $\mathbf{b} (\rightarrow \mathbf{14b})$ | CHCl ₃ , r.t. | 15b/16b | $2-(MeOCH_2)C_4H_7N$ | 83 | 85:15 | 84:16 | 73:27 ^b) | 85:15 | 68 |
| b (→14b) | CHCl ₃ , -70° | 15b/16b | $2-(MeOCH_2)C_4H_7N$ | Ι | 88:12 | Ι | Ι | 89:11 | 76 |
| c (→14c) | CHCl ₃ /MeOH. r.t. | 15c/16c | $2 - (PhNHCH_2)C_4H_7N$ | 62 | 80:20 | 82:18 | I | 80:20 | 6 |
| $\mathbf{d} (\rightarrow \mathbf{14d})$ | CHCl ₃ , r.t. | 15d/16d | $2-(MeOOC)C_4H_7N$ | 86 | 80:20 | 77:23 | I | I | 54 |
| 4 (→14e) | CHCl ₃ , r.t. | 15e/16e | $1-(t-BuOOC)C_4N_7N$ | 81 | 60:40 | 60:40 | 63:37°) | I | 20 |
| 5a (→14f) | CHCl ₃ , r.t. | 15f/16f | PhCH(OH) | 67 | 31:69 | 29:71 | I | I | 42 |
| a (→14f) | $CHCl_3, 0^\circ$ | 15f/16f | PhCH(OH) | I | 24:76 | I | Ι | I | 52 |
| $b \; (\rightarrow 14g)$ | CHCl ₃ , r.t. | 15g/16g | PhCH(OMe) | 67 | 41:59 | 42:58 | I | ļ | 16 |
| ^a) With calib | stion; column Spherison | rb; McOH/H2O | -1- | | | | | | |
| ^b) Without c | alibration; column Lichr | ospher; MeOH/H | I ₂ O 6:4. | | | | | | |
| ^c) Without c | alibration; column Spher | isorb; MeOH/M | eCN/H2O 40:6:54. | | | | | | |

Helvetica Chimica Acta – Vol. 75 (1992)



known chiral alkoxyamine (+)-1 [8] with the appropriate optically active asymmetric inductors, *i.e.* L-proline or D-mandelic-acid derivatives, led to the compounds 15 having the (1*R*,4*S*)-configuration in the bicyclic moiety (*Scheme 3*). Thus, (+)-1 reacted with phenyl chloroformate in the presence of K_2CO_3 (\rightarrow 17) and then with prolinol (6a) in pyridine to a compound which was identical with adduct 15a. Compound 15b was obtained by reaction of (+)-1 with phosgene in the presence of Et₃N (\rightarrow 18 [19]), followed by reaction with 6b in the presence of K_2CO_3 . Acylation of (+)-1 with carbamoyl chloride 8 gave 15d; stereoisomer 16d was formed as a minor product (*ca.* 8%) in this experiment, probably as a result of partial epimerization of 8. Compound 15e was obtained by direct coupling of (+)-1 with *N*-[(*tert*-butoxy)carbonyl]-L-proline (12) the dehydrating agent being dicyclohexylcarbodiimide (DCC). Identification of all these chiral coupling products with the major cycloadducts 15 was performed by ¹H- and ¹³C-NMR spectroscopy and in some cases by comparison of their optical rotations and their melting points (see *Table 2*).

Product 15f was obtained by direct coupling of (+)-1 with mandelic acid 13a in the presence of DCC, according to [20], and compound 15g resulted from coupling of (+)-1 with the acyl chloride of 13b. These two 'coupling' products – having the (R)-configura-

| Series | Adducts of | the Diels-Alder read | ction | | Independen | tly synthesized 15 |
|--------|------------|-----------------------------------|----------|--------------------------|------------|-----------------------------------|
| | 15 | | 16 | | M.p. | $[\alpha]_D$ (CHCl ₃) |
| | M.p. | $[\alpha]_D$ (CHCl ₃) | M.p. | $[\alpha]_{D}(CHCl_{3})$ | | |
| a | 80–83° | +2.4 (c=0.8) | 122–123° | -125 (c=0.6) | 76–77° | +3.0 (c=1.0) |
| b | liq. | -29 (c=1.0) | liq. | -123.4 (c=0.5) | liq. | -27.4(c=0.7) |
| c | 115–116° | -6 (c=0.8) | 132–133° | -46 (c=0.6) | _ | |
| d | | _ | - | - | liq. | +18 (c=0.7) |
| e | 114–116° | -65 (c=0.7) | 159–161° | +18 (c=0.5) | 112-114° | -65(c=1.0) |
| f | - | | 90–91° | +33 (c=1.0) | liq. | -33 (c=1.0) |
| g | - | | | _ | 103–104° | -48.8 (c=1.0) |

Table 2. Physical Data of the Adducts 15a g and 16a-g

tion in the mandelic-acid moiety and the (1R,4S)-configuration in the bicyclic moiety – are identical with the *minor* cycloadducts **15f** and **15g** which were obtained by asymmetric *Diels-Alder* cycloaddition. The absolute configuration of (+)-1 having been controversial for some time [9], opposite configurations were attributed to these adducts [14] [15].

Mechanistic Interpretation. – The absolute configuration of the major *Diels-Alder* adducts being now firmly established, a stereochemical course can be postulated for their formation. The cycloadducts obtained from the *N*-(nitrosocarbonyl) dienophiles in the *L*-proline series are urea derivatives with a planar topology, for which two pairs of rotamers should be considered: *a*) the two planar rotamers having s-*trans*- and s-*cis*-conformation at the N(1)–CO bond, the s-*cis*-rotamer (N=O *cis* to the asymmetric C(2) centre) playing obviously the dominant role during the asymmetric cycloaddition step, and *b*) the s-*trans*- and the s-*cis*-rotamers relative to the CO–NO moiety. An assumed '*endo*' approach from the least hindered side during the *Diels-Alder* cycloaddition leads to



a well-defined transition state in which the acylnitroso moiety is in its s-*cis*-conformation (see A, *Scheme 4*). Similar conclusions were reached by *Waldmann* and *Dräger* for some homo-*Diels-Alder* cycloadditions [21] as well as by *Gouverneur* and *Ghosez* [11] and by us [12] [13] for similar hetero-*Diels-Alder* cycloadditions.

In the mandelic acid series, there is a high degree of free rotation around the σ bond which connects the phenyl moiety to the chiral (*R*)-configurated C-atom and around the σ bond which connects the chiral inductor to the acylnitroso moiety. Due to the stiffness of the molecular framework of **14f** (intramolecular H-bond), two transition states **B** and **C** can be postulated (*'endo'*-approach assumed), both of which lead to cycloadduct **16f** as the major product (*Scheme 4*). **B** exhibits a H-bond between OH–C(2) and the carbonyl function which requires s-*cis*-conformation, and **C** shows a H-bond between OH–C(2) and the nitroso function which requires s-*trans*-conformation.

Transition state **B** is similar to the one described for L-proline. Such a transition state was already postulated by *Masamune* for some homo-*Diels-Alder* reactions [22]. Transition state **C** was postulated by *Kirby* and *Nazeer* [14] as well as by *Procter* and coworkers who based their assumption on quantum-chemical considerations [15] [23]. Furthermore, an interpretation for the diastereoselectivity in the *O*-methylmandelic-acid series was proposed [15] [23].

The absolute configuration of the major adducts **15a–d** and **16f** can tentatively be explained in terms of FMO theory. In the transition state the MO interactions appear to be strongest when both the diene and the acylnitroso dienophile (a heterodiene) are in their s-cis-conformation. The interaction between the HOMO of cyclohexadiene and the LUMO of the acylnitroso diene in such a transition state comprises two bonding interactions (*Diels-Alder* cycloaddition) and two secondary orbital interactions. The reasoning may also be reversed: since there are two bonding interactions and two secondary orbital interactions, cyclohexadiene and the dienophile must both be in their s-cis-conformation.

Experimental Part

General. Flash chromatography (FC): silica gel (Merck 60, 230-400 mesh).TLC: Al roll silica gel (Merck 60 F_{254}). HPLC: Spectra-Physics SP 3500 B liquid chromatograph with spectrophotometric detector SP 770; column Spherisorb Brownlee C_{18} or Lichrospher Merck C_{18e} . M.p.: Kofler hot bench or Büchi SMP 20 apparatus; corrected. [α]_D: Perkin-Elmer-PE-241 polarimeter. IR spectra (cm⁻¹): Perkin-Elmer 157 G. ¹H- and ¹³C-NMR spectra: Varian T-60, Bruker WP-80-DS, AC-F-250 and VM-400 apparatus using double-irradiation techniques; tetra-methylsilane (TMS; ¹H-NMR) and CDCl₃ or C₆D₆ (δ = 77.0 and 128.0, resp., rel. to TMS; ¹³C-NMR) as internal references; δ in ppm and J in Hz. High resolution (HR)MS: MAT-311 spectrometer; measured at the University of Rennes. Microanalyses were carried out by the Service Central de Microanalyses of the CNRS, F-69390 Vernaison.

Starting Materials. (–)-D-Mandelic acid, O-benzylhydroxylamine, methyl chloroformate, phenyl chloroformate, cyclohexa-1,3-diene, L-Boc-proline (12), tetrapropylammonium periodate were purchased from *Fluka*. Methyl N-(chlorocarbonyl)-L-prolinate (8) was purchased from *Aldrich*, phenyl carbonate from *Schering-Kahlbaum*, (2S)-2-(Anilinomethyl)pyrrolidine (6c) and (2S)-2-(Methoxymethyl)pyrrolidine (6b) from *Merck-Schuchardt*, NH₂OH·HCl and K₂CO₃ from *Prolabo*. Methyl D-mandelate was prepared by reaction of mandelic acid with 10% HCl in MeOH. L-Prolinol (6a) was prepared according to [24]. Et₃N was distilled and then kept under Ar in the presence of 4 Å molecular sicves. The usual solvents were freshly distilled. The chlorinated ones were kept over Na₂CO₃. Bicyclic hydroxylamine (+)-1 was prepared according to [8] [12].

(-)-(R)-2-Methoxy-2-phenylacetohydroxamic acid (**5b**) was prepared according to [4b]. Colourless crystals (68%). M.p. 143–144°. $[\alpha]_D^{24} = -71$ (MeOH, c = 0.5), $[\alpha]_D^{24} = -58.5$ (acetone, c = 0.6); [4b]: m.p. 137–139°; $[\alpha]_D^{23} = -63$ (acetone, c = 0.66).

(-)-(S)-2-(Hydroxymethyl)pyrrolidine-N-carbohydroxamic Acid (**3a**). To a stirred soln. of **6a** (1.15 g, 11 mmol) in pyridine (12 ml) was added hydroxamic acid 7 (1.7 g, 11 mmol). After 16 h at 30° under Ar, the soln. was evaporated and the residue purified by FC (AcOEt/MeOH 8:2): **3a** (0.92 g, 51%). Colourless crystals. M.p. 144 145° (AcOEt/i-PrOH 1:1). [α]₂₀²⁰ = -78 (MeOH, *c* = 1.5). IR (KBr): 3400, 3220, 1645, 1530, 1400, 1355, 1075, 1050, 1035, 1010, 875, 865, 775. ¹H-NMR (60 MHz, CDCl₃/CD₃OD 9:1): 4.40 (br. *s*, NHOH, OH); 3.90 (*m*, H–C(2)); 3.43 (*m*, CH₂–C(2), 2 H–C(5)); 1.83 (*m*, 2 H–C(3), 2 H–C(4)). Anal. calc. for C₆H₁₂N₂O₃ (160.17): C 44.99, H 7.55, N 17.49; found: C 44.9, H 7.5, N 17.7.

(-)-(S)-2-(Methoxymethyl)pyrrolidine-N-carbohydroxanic Acid (**3b**). To a stirred soln. of **6b** (0.417 g, 3.6 mmol) in pyridine (5 ml) was added 7 (0.593 g, 3.9 mmol, 1.1 equiv.). After 16 h under Ar at r.t., the soln. was evaporated and the phenol distilled off at 40°/0.5 Torr. The crude residue was washed with AcOEt or with Et₂O and the washing soln. separated by FC (AcOEt): **3b** (0.461 g, 72%). Colourless crystals. M.p. 91–92° (AcOEt). $[\alpha]_{D}^{20} = -92.6$ (MeOH, c = 1.0). IR (KBr): 3220, 2940, 2895, 1658, 1460, 1380, 1355, 1090, 767, 656. ¹H-NMR (60 MHz, CDCl₃): 8.25, 7.15 (2s, NHOH); 3.95 (m, H–C(2)); 3.38 (s, MeO); 3.2–3.6 (m, CH₂–C(2), 2 H–C(5)); 2.0–1.6 (m, 2 H–C(3), 2 H–C(4)). Anal. cale. for C₇H₁₄N₂O₃ (174.20): C 48.26, H 8.10, N 16.08; found: C 48.2, H 8.0, N 16.1.

(-)-(S)-2- $(Anilinomethyl)pyrrolidine-N-carbohydroxamic Acid (3c). To a stirred soln. of 6c (276 mg, 1.56 mmol) in pyridine (3 ml) was added 7 (390 mg, 2.55 mmol, 1.6 equiv.). After 16 h at 30° under Ar, pyridine was evaporated and the residue purified by FC (AcOEt): 3c (310 mg, 85%). Colourless crystals. M.p. 123 124° (AcOEt). <math>[\alpha]_D^{20} = -37$ (CHCl₃, c = 2.1). IR (KBr): 3330, 3210, 2960, 2910, 1665, 1600, 1500, 1415, 1250, 760, 700. ¹H-NMR (60 MHz, CDCl₃): 7.15 (*m*, 3 arom. H): 6.69 (*m*, 2 H_o); 4.16 (*m*, H–C(2)); 3.27 (*m*, 2 H–C(5), CH₂–C(2)); 1.95 (*m*, 2 H–C(3), 2 H–C(4)). Anal. calc. for C₁₂H₁₇N₃O₂ (235.28): C 61.25, H 7.28, N 17.86; found: C 61.4, H 7.4, N 18.0.

Methyl-1-[(Hydroxyamino) carbonyl]-L-*prolinate* (**3d**). A stirred soln. of **9** (2 g, 7.2 mmol) in AcOEt (160 ml) was submitted to hydrogenolysis over 5% Pd/C (0.30 g) at 1 atm overnight. After separation of the catalyst by centrifugation, the solvent was evaporated and the crude residue washed with Et₂O: **3d** (1.27 g, 94%). Colourless crystals. M.p. 120° (Me₂CO₃). [α]_D²⁰ = -8.4 (CHCl₃, *c* = 0.4). IR (KBr): 3360, 3170, 1720, 1665, 1645, 1460, 1435, 1390, 1375, 1230, 1210. ¹H-NMR (80 MHz, CDCl₃): 6.25 (br. *s*, NHOH); 4.48 (*t*, *J* = 5.6, H–C(2)); 3.75 (*s*, MeO); 3.46 (*t*, *J* = 6.2, 2 H–C(5)); 2.10 (*m*, 2 H–C(3), 2 H–C(4)). Anal. calc. for C₇H₁₂N₂O₄ (188.18): C 44.68, H 6.43, N 14.88; found: C 44.5, H 6.5, N 15.1.

(-)-(S)- tert-*Butyl* 2-[(Hydroxyamino)carbonyl]pyrrolidine-1-carboxylate (4). To a stirred soln. of**12**(1.02 g, 4.7 mmol) and Et₃N (0.655 ml, 4.7 mmol, I equiv.) in CCl₄ (5 mol) at 0° under Ar was added dropwise methyl chloroformate (0.360 ml, 4.7 mmol, 1 equiv.). After 16 h at r.t., the mixture was filtered, the solid material washed with CCl₄, and the combined org. soln. evaporated. The resulting mixed anhydride was dissolved under Ar in AcOEt (18 ml) and H₂O (0.45 ml) to which NH₂OH · HCl (0.409 g, 5.88 mmol, 1.25 equiv.) and K₂CO₃ (0.72 g, 5.21 mmol, 1.1 equiv.) were added. After 16 h at 25°, the mixture was filtered and the solid residue washed swends with hot AcOEt. The combined org. layer was evaporated and the solid residue washed with i-Pr₂O:**4** $(0.794 g, 73%). M.p. 167–168° (AcOEt). Colourless crystals. <math>[\alpha]_D^{17} = -51$ (MeOH, c = 0.9). IR (KBr): 3210, 3030, 2980, 2910, 1665, 1545, 1415, 1370, 1160, 1130, 1050. ¹H-NMR (60 MHz, CDCl₃): 6.80 (br. *s*, NHOH); 4.20 (*m*, H–C(2)); 3.40 (*m*, 2 H–C(5)); 2.25–1.75 (*m*, 2 H–C(3), 2 H–C(4)); 1.46 (*s*, *t*-Bu). Anal. calc. for C₁₀H₁₈N₂O₄ (230.26): C 52.16, H 7.88, N 12.17; found: C 52.3, H 8.0, N 12.1.

(-)-(R)-2-Hydroxy-2-phenylacetohydroxamic Acid (**5a**). To a stirred soln. of NH₂OH · HCl (0.70 g, 10 mmol, 2 equiv.) in MeOH (5 ml) kept at r.t. under Ar was added a soln. of KOH (0.92 g, 16.4 mmol, 3.4 equiv.) in MeOH (2.5 ml). To this mixture cooled to 0° was added methyl p-mandelate (0.80 g, 4.84 mmol). After filtration of the precipitate, the soln. was kept at 30° overnight and evaporated to 30% of its initial volume. H₂O (7 ml) and conc. HCl soln. were added to pH 6. The resulting soln. was extracted with AcOEt (6 times) and the combined org. soln. washed with brine, dried (MgSO₄), and evaporated: **5a** (0.435 g, 53%). Colourless crystals. M.p. 157–158° (dec., AcOEt; [14]: m.p. 137–138° for the (+)-(S)-isomer). [α]_D²⁰ = -46 (MeOH, c = 0.6) ([14]: [α]_D²⁰ = -164 (H₂O, c = 2.5) for the (+)-(S)-isomer!). IR (KBr): 3420, 3190, 2860, 1660, 1570, 1455, 1350, 1030, 970, 745, 700, 620. ¹H-NMR (60 MHz, CDCl₃/CD₃OD): 7.36 (m, Ph); 5.07 (s, H-C(2)); 4.26 (br. s, OH, NHOH). Anal. calc. for C₈H₉NO₃ (167.16): C 57.48, H 5.43, N 8.38; found: C 57.5, H 5.5, N 8.5.

Phenyl N-*Hydroxycarbamate* (7). a) To a stirred mixture of NH₂OH \cdot HCl (0.63 g, 9 mmol, 1.4 equiv.) and K₂CO₃ (1.06 g, 7.7 mmol, 1.2 equiv.) in Et₂O (6 ml) and H₂O (0.1 ml) kept at 0° under Ar was added dropwise phenyl chloroformate (0.8 ml, 1.0 g, 6.4 mmol). After 16 h at r.t., the mixture was filtered, the org. soln. evaporated, and the crystalline residue 7 (0.9 g, 90%) washed with benzene.

b) To a stirred soln. of phenyl carbonate (8.2 g, 40 mmol) in MeOH (30 ml) at r.t. under Ar were added $NH_2OH \cdot HCl$ (3.5 g, 50 mmol, 1.25 equiv.) and K_2CO_3 (3.5 g, 25 mmol, 1.25 equiv.). After 1 night, the mixture was

neutralized with conc. HCl soln. (0.8 ml) and filtered, the org. soln. evaporated, and the phenol distilled at $45^{\circ}/0.5$ Torr. The solid residue was washed with C₆H₆: 7 (3.63 g, 62%). Colourless dimorphous crystals. M.p. $105-107^{\circ}$ (C₆H₆; [17]: m.p. $105-107^{\circ}$). IR (KBr, form I): 3400, 3240, 1700, 1525, 1480, 1270, 1200, 1095, 1020, 790, 685. IR (KBr, form 11): 3280, 1680, 1510, 1480, 1285, 1205, 1160, 1100, 790, 710, 685. ¹H-NMR (60 MHz, CDCl₃): 8.33 (*s*, NH); 7.20 (*m*, Ph); 3.40 (br. *s*, OH).

(-)-(S)-*Methyl* 1-{ $[(Benzyloxy)amino]carbonyl}pyrrolidine-2-carboxylate (9). To a stirred soln. of 8 (0.93 g, 4.8 mmol) in AcOEt (10 ml) were added at r.t. under Ar NH₂OBn·HCl (0.86 g, 5.4 mmol, 1.1 equiv.), K₂CO₃ (0.715 g, 5.2 mmol, 1.08 equiv.), and H₂O (0.2 ml). This mixture was left at 60° for 40 h and then filtered, the precipitate washed with AcOEt and CH₂Cl₂, the combined org. layer evaporated, and the crystalline residue washed with cyclohexane: 9 (1.23 g, 91%). M.p. 123–124° (AcOEt/cyclohexane 2:1). [<math>\alpha$]_D²⁰ = -63 (CHCl₃, c = 1.3). IR (KBr): 3230, 2980, 2960, 1735, 1650, 1500, 1385, 1365, 1200, 1170, 1025, 745, 700. ¹H-NMR (80 MHz, CDCl₃): 7.39 (*m*, C₆H₅); 4.87 (*s*, PhCH₂); 4.44 (*t*, J = 5.4, H–C(2)); 3.71 (*s*, MeO); 3.41 (*dt*, J = 1.0, 6.2, 2 H–C(5)); 2.0 (*m*, 2 H–C(3), 2 H–C(4)). Anal. calc. for C₁₄H₁₈N₂O₄ (278.31): C 60.42, H 6.52, N 10.07; found: C 60.3, H 6.6, N 10.3.

(5 S)-3-(Benzyloxy)-1,3-diazabicyclo[3.3.0]octane-2,4-dione (10). When left to stand over 40 h at 60° or on column chromatography, **9** cyclized quantitatively to **10**. Colourless crystals. M.p. 99–100° (i-PrOH). [α]_D²⁰ = -79 (CHCl₃, c = 0.5). IR (KBr): 2960, 2900, 1785, 1715, 1405, 1310, 1235, 1175, 965, 940, 915, 740, 700. ¹H-NMR (80 MHz, CDCl₃): 7.38 (m, Ph); 5.14 (s, PhCH₂); 3.93 (dd, J = 7.6, 8.9, H-C(5)); 3.59 (dt, $J = 11.1, 7.1, 7.2, \text{H}_{a}-\text{C}(8)$); 3.23 (m, $J = 11.3, 5.5, 6.8, \text{H}_{b}-\text{C}(8)$); 2.45–1.35 (m, 2 H–C(6), 2 H–C(7)). ¹³C-NMR (20.1 MHz, CDCl₃): 168.2 (s, C(4)); 156.3 (s, C(2)); 133.1 (s, C_{ipso}); 129.7 (d, $J = 162, \text{C}_{o}$); 129.0 (d, $J = 162, \text{C}_{p}$); 128.1 (d, $J = 161, \text{C}_{m}$); 78.6 (t, $J = 150, \text{PhCH}_2\text{O}$); 60.4 (d, J = 152, C(5)); 45.3 (t, J = 145, C(8)); 27.0 (t, J = 136, C(6)); 25.9 (t, J = 134, C(7)). Anal. cale. for C₁₃H₁₄N₂O₃ (246.27): C 63.40, H 5.73, N 11.38; found: C 63.4, H 5.8, N 11.5.

(5S)-3-Hydroxy-1,3-diazabicyclo[3.3.0]octane-2,4-dione (11). a) A stirred soln. of 10 (0.50 g, 2 mmol) in AcOEt (30 ml) containing 5% Pd/C (84 mg) was put under H₂ (1 atm) for 4 h at 30°. Centrifugation and rinsing of the catalyst with AcOEt/CH₂Cl₂ gave 11 (0.312 g, 98%). Colourless crystals.

b) Reaction of **8** with NH₂OH · HCl and K₂CO₃, followed by addition of HCl/MeOH, gave **11** quantitatively. M.p. 117–118° (AcOEt/cyclohexane). $[\alpha]_D^{20} = -9.6$ (CHCl₃, c = 0.4). ¹H-NMR (80 MHz, CDCl₃): 5.85 (br. *s*, NOH); 4.09 (*t*, J = 8.2, H–C(5)); 3.58 (*dt*, J = 7.2, 7.6, 11.2, H_a–C(8)); 3.29 (*ddd*, J = 4.8, 6.9, 11.2, H_b–C(8)); 2.50–1.60 (*m*, 2 H–C(6), 2 H–C(7)). ¹³C-NMR (20.1 MHz, CDCl₃): 170 (*s*, C(4)); 158.1 (*s*, C(2)); 60.8 (*d*, J = 152, C(5)); 45.3 (*t*, J = 145, C(8)); 26.4 (*t*, J = 136, C(6) or C(7)); 26.3 (*t*, J = 134, C(7) or C(6)). Anal. calc. for C₆H₈N₂O₃ (156.14): C 46.15, H 5.16, N 17.94; found: C 46.1, H 5.2, N 18.0.

General. Procedure for the Diels-Alder Cycloaddition with Cyclohexa-1,3-diene. To a stirred soln. of cyclohexa-1,3-diene and $(Pr_4N)IO_4$ (0.33 mmol per mol of diene) in CHCl₃ (1–3 ml per mmol) containing *ca*. 10 beads of 4 Å molecular sicves was added portionwise within 30–45 min the hydroxamic acid; if necessary, some MeOH was added. After 1 h at r.t., the mixture was diluted with Et₂O and washed with 1M NaHCO₃ (1 ml) containing some Na₂SO₃ until decoloration and with brine. The aq. phase was extracted several times with Et₂O and the combined Et₂O soln. dried (MgSO₄) and evaporated.

(1R,4S)-3-[(2S)-2-(Hydroxymethyl)pyrrolidine-1-carbonyl]-2-oxa-3-azabicyclo[2.2.2]oct-5-ene (15a) and Its (1S,4R)-Diastereoisomer 16a. From cyclohexa-1,3-diene (0.142 ml, 1.5 mmol), (Pr₄N)IO₄ (0.188 g, 0.5 mmol), and 3a (0.237 g, 1.5 mmol) in CHCl₃ (3 ml) and MeOH (1 ml). The oily residue (0.315 g, 89%) was separated by FC (AcOEt): 15a and 16a.

Major Adduct **15a**: Colourless crystals (147 mg, 42%). M.p. $80-83^{\circ}$ (AcOEt/(i-Pr)₂O 1:6). $[\alpha]_D^{20} = +2.4$ (CHCl₃, c = 0.8). IR (KBr): 3370, 2960, 1650, 1620, 1415, 1370, 1205, 1090, 1058, 920, 768, 710. ¹H-NMR (400 MHz, C₆D₆): *Tables 3* and 4. ¹H-NMR (250 MHz, CDCl₃): 6.63 (*ddd*, H–C(5)); 6.52 (*ddd*, H–C(6)); 4.63 (*m*, H–C(1), H–C(4)); 4.21 (*m*, OH); 4.10 (*m*, H–C(2')); 3.75 (*m*, H_a–C(5')); 3.61 (*m*, H₂C(α)); 3.43 (*m*, H_b–C(5')); 2.16 (*m*, H_a–C(7), H_a–C(8)); 1.96 (*m*, H_a–C(3'), H_b–C(4')); 1.78 (*m*, H_b–C(3'), H_b–C(4')); 1.50 (*m*, H_b–C(7), H_b–C(8)). ¹³C-NMR (20.1 MHz, CDCl₃): *Table 5*. Anal. calc. for C₁₂H₁₈N₂O₃ (238.29): C 60.48, H 7.61, N 11.76; found: C 60.2, H 7.6, N 11.8.

Minor Adduct **16a**: Colourless crystals (49 mg, 14%). M.p. 122–123° (AcOEt/(i-Pr)₂O). [α]_D²⁰ = -125 (CHCl₃, c = 0.6). IR (KBr): 3430, 2980, 2940, 2890, 1620, 1420, 1390, 1375, 1195, 1050, 900, 880, 835, 778, 710. ¹H-NMR (400 MHz, C₆D₆): *Tables 3* and 4. ¹³C-NMR (20.1 MHz, CDCl₃): *Table 5*. Anal. calc. for C₁₂H₁₈N₂O₃ (238.29): C 60.48, H 7.61, N 11.76; found: C 60.2, H 7.6, N 11.9.

(1R,4S)-3-[(2S)-2-(Methoxymethyl)pyrrolidine-1-carbonyl]-2-oxa-3-azabicyclo[2.2.2]oct-5-ene (15b) and Its (1S,4R)-Diastereoisomer 16b. From cyclohexa-1,3-diene (37 µl, 0.39 mmol), (Pr₄N)IO₄ (60 mg, 0.16 mmol), and 3b (67 mg, 0.38 mmol) in CHCl₃ (1 ml). The crude oily residue (117 mg) was separated by prep. TLC (AcOEt): 15b and 16b.

Major Adduct **15b**: Colourless oil (60 mg, 62%). $[\alpha]_{D}^{20} = -29$ (CHCl₃, c = 1.0). IR (CCl₄): 2970, 2940, 2890, 1640, 1405, 1365, 1195, 1110, 900, 875, 700. ¹H-NMR (250 MHz, C₆D₆): *Tables 3* and 4. ¹H-NMR (80 MHz, CDCl₃): 6.56 (*m*, H–C(5), H–C(6)); 4.61 (*m*, H–C(1), H–C(4)); 4.13 (*m*, H–C(2')); 3.51 (*dd*, H_a–C(α)); 3.50 (*m*, 2 H–C(5')); 3.34 (*s*, MeO); 3.29 (*dd*, H_b–C(α)); 2.30–1.10 (*m*, 2 H–C(7), 2 H–C(8), 2 H–C(3'), 2 H–C(4')); ¹³C-NMR (20.1 MHz, CDCl₃): *Table 5*. MS: 252 (5), 235 (3), 142 (69), 114 (22), 82 (53), 80 (26), 79 (51), 77 (22), 71 (20), 70 (100), 68 (17), 67 (45). HR-MS: 252.1489 (C₁₃H₂₀N₂O₃, *M*⁺, calc. 252.14738).

Minor Adduct **16b**: Colourless oil (11 mg, 11%). [α]_D²⁰ = -123.4 (CHCl₃, *c* = 0.5). IR (CCl₄): 2970, 2940, 2890, 1645, 1410, 1375, 1120, 880. ¹H-NMR (80 MHz, C₆D₆): *Tables 3* and 4. ¹H-NMR (80 MHz, CDCl₃): 6.67 (*ddd*, H–C(5)); 6.47 (*ddd*, H–C(6)); 4.72 (*m*, H–C(4)); 4.56 (*m*, H–C(1)); 4.18 (*m*, H–C(2')); 3.53 (*m*, 2 H, H–C(5')); 3.50 (*dd*, H_a–C(α)); 3.33 (*s*, MeO); 3.28 (*dd*, H_b–C(α)); 2.20 1.20 (*m*, 2 H–C(7), 2 H–C(8), 2 H–C(3'), 2 H–C(4')). ¹³C-NMR (20.1 MHz, CDCl₃): *Table 5*. MS: 252 (13), 142 (100), 129 (28), 114 (34), 110 (18), 82 (88), 80 (23), 79 (37), 70 (64), 69 (13), 68 (16), 67 (28). HR-MS: 252.1477 (C₁₃H₂₀N₂O₃, *M*⁺, calc. 252.14738).

(1R,4S)-3-[(2S)-2-(Anilinomethyl)pyrrolidine-1-carbonyl]-2-oxa-3-azabicyclo[2.2.2]oct-5-ene (15c) and Its (1S,4R)-Diastereoisomer 16c. From Cyclohexa-1,3-diene (0.102 ml, 1.07 mmol), (Pr₄N)IO₄ (0.146 g, 0.39 mmol), and 3c (0.253 g, 1.07 mmol) in CHCl₃ (3 ml) and MeOH (1 ml). The mixture was separated by prep. TLC (AcOEt/cyclohexane 4:6): 15c and 16c.

Major Adduct **15c**: Colourless crystals (0.222 g, 65%). M.p. 115–116° (AcOEt/(i-Pr)₂O 1:4). $[x]_{D}^{20} = -6$ (CHCl₃, c = 0.8). IR (KBr): 3370, 2960, 2940, 2900, 1660, 1600, 1520, 1390, 1365, 1325, 1210, 915, 750, 695. ¹H-NMR (400 MHz, C₆D₆): *Tables 3* and 4. ¹³C-NMR (20.1 MHz, CDCl₃): *Table 5*. Anal. calc. for C₁₈H₂₃N₃O₂ (313.40): C 69.98, H 7.40, N 13.41; found: C 69.0, H 7.4, N 13.3.

Minor Adduct **16c**: Colourless crystals (46 mg, 14%). M.p. 132–133° (AcOEt/(i-Pr)₂O 1:2.5). $[\alpha]_{D}^{20} = -46$ (CHCl₃, c = 0.6). IR (KBr): 3360, 3060, 2970, 2940, 1635, 1605, 1520, 1500, 1485, 1415, 1370, 1315, 880, 760, 695. ¹H-NMR (400 MHz, C₆D₆): *Tables 3* and 4. ¹³C-NMR (20.1 MHz, CDCl₃): *Table 5*. Anal. calc. for C₁₈H₂₃N₃O₂ (313.40): C 68.98, H 7.40, N 13.41; found: C 69.1, H 7.5, N 13.5.

(IR,4S)-3-[(2S)-2-(*Methoxycarbonyl*)*pyrrolidine-1-carbonyl*]-2-oxa-3-azabicyclo[2.2.2]oct-5-ene (**15**) and *Its* (*IS*,4R)-*Diastereoisomer* **16d**. From cyclohexa-1,3-diene (0.250 ml, 2.6 mmol). (Pr₄N)IO₄ (0.335 g, 0.9 mmol), and **3d** (0.497 g, 2.6 mmol) in CHCl₃ (3 ml). The mixture was purified but not separated by FC (AcOEt/cyclohexane 8:2) leading to an oily mixture (86%) of the major **15d** and the minor **16d**. [α]²_D = -7 (CHCl₃, *c* = 0.7). IR (CCl)₄: 2970, 2940, 2890, 1750, 1650, 1405, 1365, 1195, 1165, 875. ¹H-NMR (250 MHz, C₆D₆): *Tables 3* and 4. ¹H-NMR (80 MHz, CDCl₃, **15d**): 6.69 (*m*, H–C(5)); 6.45 (*m*, H–C(6)); 4.78 (*m*, H–C(4)); 4.46 (*m*, H–C(1), H–C(2')); 3.70 (*s*, CO₂Me); 3.61 (*m*, 2 H–C(5')); 2.34–1.16 (*m*, 2 H–C(7), 2 H–C(8), 2 H–C(3'), 2 H–C(4')). ¹³C-NMR (20.1 MHz, CDCl₃): *Table 5*. MS: 266 (17), 156 (15), 129 (37), 128 (100), 80 (17), 79 (34), 77 (12), 70 (25), 68 (11). HR-MS: 266.1269 (C₁₃H₁₈N₂O₄, *M*⁺, calc. 266.12665).

 $(1R,4S)-3-\{(2S)-1-[(tert-Butoxy)carbonyl]pyrrolidine-2-carbonyl]-2-oxa-3-azabicyclo[2.2.2]oct-5-ene (15e) and Its (1S,4R)-Diastereoisomer 16e. From cyclohexa-1,3-diene (0.165 ml, 1.72 mmol), (Pr₄N)IO₄ (0.23 g, 0.61 mmol), and 4 (0.397 g, 1.72 mmol) in CHCl₃ (4 ml). The colourless solid residue was purified by FC (0.431 g, 81%) and separated on a Jobin-Yvon steel column filled with silicic acid (AcOEt/MeOH 99:1) under N₂ pressure (4 atm): 15e and 16e.$

Major Adduct **15e**: Colourless crystals. M.p. 114–116° ((i-Pr)₂O). $[\alpha]_D^{20} = -65$ (CHCl₃, c = 0.7). IR (KBr): 2980, 2950, 2880, 1695, 1660, 1620, 1405, 1370, 1170, 1135, 1085, 962, 915, 702. ¹H-NMR (400 MHz, CD₂Cl₂): *Tables 3* and 4. ¹³C-NMR (20.1 MHz, CDCl₃): *Table 5*. Anal. cale. for C₁₆H₂₄N₂O₄ (308.38): C 62.31, H 7.85, N 9.09; found: C 62.0, H 7.9, N 9.1.

Minor Adduct **16e**: Colourless crystals. M.p. 159–161° (AcOEt/(i-Pr)₂O 1:3). $[\alpha]_D^{20} = +18$ (CHCl₃, c = 0.5). IR (KBr): 2990, 2970, 2940, 2890, 1690, 1645, 1405, 1370, 1180, 1170, 1130, 1085, 965, 915, 775, 720. ¹H-NMR (400 MHz, CD₂Cl₂): *Tables 3* and *4*. ¹³C-NMR (20.1 MHz, CDCl₃): *Table 5*. Anal. calc. for C₁₆H₂₄N₂O₄ (308.38): C 62.31, H 7.85, N 9.09; found: C 62.3, H 8.0, N 9.2.

(1R,4S)-3-[(2R)-2-Hydroxy-2-phenylacetyl]-2-oxa-3-azabicyclo[2.2.2]oct-5-ene (15f) and Its (1S,4R)-Diastereoisomer 16f. From cyclohexa-1,3-diene (0.135 ml, 1.42 mmol), (Pr₄N)IO₄ (0.182 g, 0.4 mmol), and 5a (0.237 g, 1.42 mmol) in CHCl₃ (3 ml). The yellowish solid residue was purified by prep. TLC (AcOEt) without separation of the two adducts: 0.232 g, 67%. The major adduct was obtained pure by fractional crystallization ((i-Pr)₂O).

Major Adduct **15**f: Colourless crystals. M.p. 90–91° (((i-Pr)₂O). $[\alpha]_{D}^{24} = +33$ (CHCl₃, c = 1.0). IR (KBr): 3420, 3060, 2980, 2940, 1630, 1610, 1390, 1360, 1275, 1190, 1170, 1090, 1065, 1045, 950, 900, 865, 835, 720, 700. ¹H-NMR (400 MHz, C₆D₆): *Tables 3* and 4. ¹H-NMR (250 MHz, CDCl₃): 7.33 (m, 2 arom. H); 7.25 (m, 3 arom. H); 6.41 (ddd, H–C(5)); 6.02 (m, H–C(6)); 5.28 (d, H–C(2')); 5.19 (m, H–C(4)); 4.56 (m, H–C(1)); 4.15 (d, OH–C(2'));

| | H-C(I |) H-C(4 | i) H-C(5 | H-C(t | 5) $H_a - C(7), H_a - C_1$ | (8) H _b -C | C(7),H _b -C(| 8) H–C(2' |) CH ₂ (3'),CH ₂ (| 4') H _a -C(: | 5'),H _b -C(5 |) H _a -C(| (x) | $H_{b}-C(\alpha)$ | Other data |
|-------------------------------|-------------------|--------------|--------------|---------------------------|----------------------------|-----------------------|-------------------------|------------------------|--|-------------------------|-------------------------|----------------------|-------------------------|---|-------------------|
| 15a | 4.14 | 4.63 | 6.47 | 6.08 | 1.88 2.00 | 0.94 | 1.07 | 4.06 | 1.15-1.45 | 3.72 | 3.30 | 3.64 | 3.56 | 4.21 (OH) | 343 K, 400 MHz |
| 16a | 4.12 | 4.71 | 6.52 | 6.05 | 1.87 | 16.0 | 1.15 | 4.20 | 1.10-1.50 | 3.68 | 3.34 | 3.65 | 3.65 | 3.57 (OH) | 343 K, 400 MHz |
| 15b | 4.17 | 4.75 | 6.52 | 6.12 | 1.91 2.06 | 0.96 | 1.11 | 4.28 | 1.35–1.85 | 3.73 | 3.40 | 3.64 | 3.37 | 3.17 (MeO) | 338 K, 400 MHz |
| 16b | 4.14 | 4.83 | 6.64 | 6.09 | 1.85 | č | - 1.0 | 4.42 | 1.10–1.75 | 3.64 | 3.54 | 3.61 | 3.36 | 3.14 (MeO) | 303 K, 80 MHz |
| 15c | 4.14 | 4.69 | 6.46 | 6.11 | 1.89 2.10 | 0.95 | 1.08 | 4.27 | 1.30-1.50 | 3.71 | 3.30 | 3.32 | 2.87 | 6.68 (H _o); 7.17 (H _m) 6.72 (H _p); 4.33 (NH) | 343 K, 400 MHz |
| 16c | 4.12 | 4.78 | 6.59 | 6.04 | 1.89 | 0.92 | 1.14 | 4.40 | 1.10-1.60 | 3.65 | 3.45 | 3.19 | 2.98 | 6.65 (H _o); 7.17 (H _m) 6.72 (H _p); 4.40 (NH) | 343 K, 400 MHz |
| 15d | 4.11 | 4.81 | 6.54 | 6.07 | 1.89 | 0.91 | 1.06 | 4.52 | 1.25–1.75 | 3.67 | 3.62 | T | I | 3.40 (CO ₂ Me) | 323 K, 250 MHz |
| 16d | 4.17 | 4.78 | 6.53 | 6.11 | n.d. | ц | | 4.69 | n.d. | n.c | Ť | I | I | 3.39 (CO ₂ Me) | 323 K, 250 MHz |
| 15e^b) | 4.73 | 5.14 5.10 | 6.61 6.60 | 6.51 6.49 | ca.2.1 ca.2.2 | 1.46 | 1.41 | 4.44 4.45 | 1.60-2.10 | 3.40 | 3.30 3.29 | | | 1.39 (<i>t</i> -Bu) 1.35 (<i>t</i> -Bu) | 278 K, 400 MHz |
| 16e ^c) | 4.74 | 5.14 | 6.61 | 6.47 6.54 | 2.01 2.14 | 1.47 | 1.77 | 4.33 4.37 | 1.70-2.20 | 3.40 | 3.29 | ł | i | 1.38 (<i>t</i> -Bu) 1.30 (<i>t</i> -Bu) | 278 K, 400 MHz |
| 15f | 3.84 | 4.97 | 5.95 | 5.71 | 1.10 1.37 | 0.5 | -0.7 | 5.42 | I | | | I | I | 7.55 (H _o); ca. 7.1 (H _m , H _p); 4.71 (OH) | 303 K, 250 MHz |
| 16f | 3.72 | 4.93 | 5.69 | 5.32 | 1.54 | 0.5 | -0.7 | 5.54 | I | I | | I | I | 7.40 (H _o); ca. 7.1 (H _m ,H _p); 4.69 (OH) | 303 K, 250 MHz |
| 15g | 4.09 | 5.08 | 6.16 | 5.92 | 1.42 1.55 | 0.72 | 0.80 | 5.19 | I | I | | I | I | 7.71 (H _o); ca. 7.1 (H _m , H _p); 3.33 (MeO) | 343 K, 400 MHz |
| 16g | 4.00 | 5.04 | 5.90 | 5.55 | 1.78 | 0.80 | 0.91 | 5.21 | I | ļ | | I | I | 7.57 (H _o); ca. 7.1 (H _m , H _p); 3.36 (MeO) | 343 K, 400 MHz |
| ^a) Prin (lower | ned nur line). | thers refi | er to the | proline | or mandelic moi | ety, C(α |) is <i>C</i> -C(| 2'). ^b) In | CD ₂ Cl ₂ ; 2 rota | mers in ra | tio 1:1. ° | In CD | 2Cl ₂ ; 2 rc | stamers in ratio 7 (upp | er line); 3 |

Helvetica Chimica Acta - Vol. 75 (1992)

119

| Image: Normal and the state of th | | 1910 | 10100 | 125 121 | 11 71 | 191 | | | עלא פרא | | | mpt fo | | | 11 - 0506 | 1141 5101 | 1147 5121 | 11510 514 |
|--|--------------------|--------|--------|---------|---------|--------|---------|--------------------|----------|--------|------------|--------|-----------|--------------|-------------|------------|------------|-------------|
| | | (c'1)r | (0'1)/ | J(1,/a) | J(1,/0) | (C,4)L | J (4,0) | J (4,0a) | J (4,0D) | (0°C)1 | ر بر ار | () | J (2, 34, | (a x, z) r (| J (2 4,2 D) | (B C, 4) L | (0 C, +) L | (0 C,B C) L |
| | 15a | 1.7 | 5.7 | 3.8 | 1.7 | 5.9 | 1.7 | 3.0 | 3.0 | 8.2 | 7.5 | 4.4 | 7.1 | 3.6 | 10.8 | 7.2 7.2 | 5.7 7.3 | 11.3 |
| | 16a | 1.6 | 5.8 | 4.0 | 1.4 | 5.4 | 1.8 | 2.9 | 2.9 | 8.3 | n.6 | Ţ. | n.d. | n.d. | n.d. | n.d. | 6.6 9.8 | 11.5 |
| | 15b | 1.6 | 5.8 | 3.9 | 1.6 | 5.9 | 1.7 | 3.1 | 3.2 | 8.2 | 7.0 | 3.0 | 3.4 | 7.5 | 9.2 | n.d. | n.d. | n.d. |
| Iscal1.55.83.81.56.01.73.43.28.23.77.05.96.412.47.17.17.15.36.711.3Ide1.55.83.91.55.71.82.92.98.37.17.16.45.412.24.07.36.98.611.3Ide1.55.81.11.55.81.73.03.08.42.97.77.16.45.412.24.07.36.98.611.3Ide1.55.64.01.55.81.73.03.08.42.97.7 $ -$ | 16b | 1.6 | 5.8 | 4.1 | 1.5 | 5.6 | 1.8 | 2.8 | 2.8 | 8.3 | 6.7 | 6.7 | 3.5 | 6.8 | 9.3 | n.d. | n.d. | n.d. |
| Ic 1.5 5.8 3.9 1.5 5.7 1.8 2.9 2.9 8.3 7.1 7.1 6.4 5.4 12.2 4.0 7.3 6.9 8.6 11.1 Isd ^b 1.6 5.8 4.1 1.5 5.8 1.7 3.0 8.4 2.9 7.7 - - - n.d. n.d. n.d. n.d. Isd ^b 1.6 5.8 4.1 1.5 5.8 1.7 3.0 3.0 8.4 2.9 7.7 - - - n.d. | 15c ^a) | 1.5 | 5.8 | 3.8 | 1.5 | 6.0 | 1.7 | 3.4 | 3.2 | 8.2 | 3.7 | 7.0 | 5.9 | 6.4 | 12.4 | 7.1 7.1 | 5.3 6.7 | 11.2 |
| | 16c | 1.5 | 5.8 | 3.9 | 1.5 | 5.7 | 1.8 | 2.9 | 2.9 | 8.3 | 7.1 | 7.1 | 6.4 | 5.4 | 12.2 | 4.0 7.3 | 6.9 8.6 | 11.5 |
| Iod 1.5 5.6 4.0 1.5 5.8 1.7 3.0 3.0 8.3 6.5 7.9 - - n.d. | 15d ^b) | 1.6 | 5.8 | 4.1 | 1.5 | 5.8 | 1.7 | 3.0 | 3.0 | 8.4 | 2.9 | 7.7 | I | I | I | n.d. | n.d. | n.d. |
| Ise 1.7 5.8 n.d. n.d. 6.2 1.7 n.d. | 16d | 1.5 | 5.6 | 4.0 | 1.5 | 5.8 | 1.7 | 3.0 | 3.0 | 8.3 | 6.5 | 7.9 | I | I | I | n.d. | n.d. | n.d. |
| Ide 1.8 5.7 n.d. n.d. 6.3 1.6 n.d. n.d. 8.2 8.7 3.4 - 7.5 5.0 7.5 7.5 10.4 Ist 1.6 5.6 3.7 1.8 6.1 1.7 3.6 2.2 8.2 7.2° - | 15e | 1.7 | 5.8 | n.d. | n.d. | 6.2 | 1.7 | n.d. | n.d. | 8.1 | 8.5 | 3.5 | I | | I | 7.2 5.0 | 7.2 7.2 | 10.3 |
| Isf 1.6 5.6 3.7 1.8 6.1 1.7 3.6 2.2 8.2 7.2 ^c - | 16e | 1.8 | 5.7 | n.d. | n.d. | 6.3 | 1.6 | n.d. | n.d. | 8.2 | 8.7 | 3.4 | i | | I | 7.5 5.0 | 7.5 7.5 | 10.4 |
| Iof 1.6 5.5 3.7 1.6 6.1 1.7 2.8 ^d 8.2 7.4° - - - - - - - I5g 1.7 5.6 3.7 1.7 6.2 1.7 3.2 2.8 8.2 - <td>15f</td> <td>1.6</td> <td>5.6</td> <td>3.7</td> <td>1.8</td> <td>6.1</td> <td>1.7</td> <td>3.6</td> <td>2.2</td> <td>8.2</td> <td>7.2</td> <td>(.</td> <td>Ι</td> <td></td> <td>I</td> <td>I</td> <td>I</td> <td>Ι</td> | 15f | 1.6 | 5.6 | 3.7 | 1.8 | 6.1 | 1.7 | 3.6 | 2.2 | 8.2 | 7.2 | (. | Ι | | I | I | I | Ι |
| ISg 1.7 5.6 3.7 1.7 6.2 1.7 3.2 2.8 8.2 - <th< td=""><td>16f</td><td>1.6</td><td>5.5</td><td>3.7</td><td>1.6</td><td>6.1</td><td>1.7</td><td>2.8^d)</td><td></td><td>8.2</td><td>7.4</td><td>(;</td><td>I</td><td></td><td>I</td><td>I</td><td>I</td><td>I</td></th<> | 16f | 1.6 | 5.5 | 3.7 | 1.6 | 6.1 | 1.7 | 2.8 ^d) | | 8.2 | 7.4 | (; | I | | I | I | I | I |
| 16g 1.6 5.6 3.8 1.6 6.1 1.7 2.8 ^d) 8.1 | 15g | 1.7 | 5.6 | 3.7 | 1.7 | 6.2 | 1.7 | 3.2 | 2.8 | 8.2 | Ι | | I | | I | 4 | I | I |
| | 16g | 1.6 | 5.6 | 3.8 | 1.6 | 6.1 | 1.7 | 2.8 ^d) | | 8.1 | Ι | | Ι | | I | I | I | I |

Helvetica Chimica Acta – Vol. 75 (1992)

| | Tablé | 5. ¹³ C-NM. | R Spectral | Data of Ada | tucts 15a-g | and 16a -g 1 | in CDCl ₃ a | <i>tt 300 K. δ</i> in | ppm, inter | nal standar | d CDCl ₃ (ð | $(CDCl_3) = 77 \text{ ppm})^{a}$ |
|--------------------------|-----------------------|---------------------------|----------------------------|------------------------|---------------------------------|---------------------------|------------------------|----------------------------------|-----------------------|---------------------------------|------------------------|---|
| | C(1) | C(4) | C(5) | C(6) | C(7) | C(8) | NCO | C(2') | C(3') | C(4′) | C(5') | Other data |
| 15a | 70.1 (J = 153) | 49.6 (J = 152) | 132.8 (J = 170) | 131.1 (J = 169) | (J = 132) | 19.8 (J = 133) | 163.0 | 61.0 (J = 143) | 27.8 (J = 132) | 23.6° ($J = 132$) | 48.5 (J = 143) | $66.2 (J = 144, C(\alpha))$ |
| 16a | 70.2 | 49.8 | 134.4 | 130.4 | 24.4°) | 19.6 | 164.4 | 61.0 | 27.7 | 24.0°) | 49.2 | 66.2 (C(x)) |
| 15b | 69.6 ($J = 153$) | 49.5 (J = 152) | 132.8 (J = 170) | 130.7 ($J = 169$) | 23.6 (J = 133) | 19.6 (J = 133) | 161.3 | <i>57.2</i> (<i>J</i> = 143) | 27.1 (J = 132) | 22.9 ($J = 133$) | 47.7 ($J = 143$) | 72.8 ($J = 141$, C(α)); 58.2 ($J = 140$, MeO) |
| 16b | 69.7 | 48.8 | 134.3 | 130.1 | 24.0 ^b) | 19.6 | 162.2 | 57.0 | 27.4 | 23.9 ^c) | 48.8 | 72.8 C(a); 58.4 (MeO) |
| 15c | 69.8 (J = 156) | 50.0 (<i>J</i> = 152) | 132.1 (<i>J</i> = 170) | 131.1 ($J = 169$) | $23.5^{\rm b}$ ($J = 132$) | 19.7 (<i>J</i> = 132) | 162.3 | 57.4 (J = 143) | 28.3 (J = 132) | 23.4° ($J = 132$) | 47.9 (J = 142) | $47.5 (J = 136, C(\alpha));$ 148.1, 128.7 (J = 158), 116.3 (J = 160), 112.0 (J = 155, arom. C) |
| 16c | 70.0 | 49.1 | 134.4 | 130.3 | 24.2 ^b) | 9.61 | 163.5 | 57.1 | 28.6 | 24.0°) | 49.1 | 47.7 (C(α)); 148.2, 128.7, 116.3, 112.0 (arom. C) |
| 15d | 70.2 ($J = 154$) | 49.3 ($J = 153$) | 134.2 ($J = 170$) | 130.4 ($J = 170$) | 23.9 ($J = 132$) | 19.4 ($J = 134$) | 161.4 | 60.1 (J = 152) | 29.6 ($J = 134$) | 22.8 ($J = 133$) | 48.1 ($J = 144$) | 173.3 (CO ₂); 51.3 ($J = 146.5$, Me) |
| 16d | 70.2 | 49.2 | 134.1 | 130.7 | 24.2 | 19.7 | 161.5 | 60.0 | 29.3 | 23.7 | 48.6 | 172.8 (CO ₂); 51.4 (Me) |
| $15e^{d}$) | 71.1 | 46.2 | 132.4 | 131.0 | 23.0 | 20.6 | 153.6 | 56.8 | 29.4 | 23.0 | 46.2 | 171.5 (CO ₂); 78.5 (Me ₃ C); 27.9 (Me) |
| 16e^d) | 71.3 | 46.2 | 132.2 | 130.9 | 23.0 | 20.6 | 153.6 | 57.4 | 29.4 | 23.0 | 46.2 | $171.3 (CO_2); 78.4 (Me_3C); 27.9 (Me)$ |
| 15f | 71.1 | 47.2 | 131.7 | 131.7 | 22.1 | 20.8 | 170.1 | 71.6 | I | ļ | 1 | 139.5, 127.9, 127.7, 127.0 (arom. C) |
| 16f | 70.9 ($J = 153$) | 47.4 ($J = 153$) | 131.5 (J = 170) | 130.8 (J = 170) | 23.1 $(J = 133)$ | 20.2 (J = 133) | 170.8 | 71.4 ($J = 147$) | i | I | 1 | 138.0, 127.7, 127.4 ($J = 160$), 127.2 ($J = 160$), (arom. C) |
| 15g | 71.7 | 46.8 | 132.3 | 131.8 | 22.6 | 20.9 | 169.2 | 80.6 | I | I | I | 57.0 (MeO); 136.7, 128.2, 128.0 (arom. C) |
| 16g | 71.5 ($J = 153$) | 47.2 (<i>J</i> = 153) | 132.3 (<i>J</i> = 170) | 130.6 (J = 169) | 23.6 (J = 133) | 20.4 ($J = 133$) | 170.4 | 80.6 (<i>J</i> = 145) | I | I | I | 57.1 (J = 142, MeO); 135.1, 128.2 (J = 160), 128.0 (J = 160), 128.0 (J = 160), 128.0 (J = 160), (arom. C) |
| ^a) Prim(| ed atoms refe | ar to the pro. | line or man | delic moiety | y. $C(\alpha)$ is C | -C(2'). ^b) | Or C(4'). | ^c) Or C(7). | ^d) 328 K. | | | |

Helvetica Chimica Acta – Vol. 75 (1992)

121

2.17 (*m*, H_a –C(7), H_a –C(8)); 1.36 (*m*, H_b –C(7), H_b –C(8)). ¹³C-NMR (20.1 MHz, CDCl₃): *Table 5*. Anal. calc. for C₁₄H₁₅NO₃ (245.28): C 68.55, H 6.16, N 5.71; found: C 68.5, H 6.1, N 5.7.

(1 R,4 S)-3-[(2 R)-2-Methoxy-2-phenylacetyl]-2-oxo-3-azabicyclo[2.2.2]oct-5-ene (**15g**) and Its (1S,4 \text{ R})-Diastereoisomer **16g**. From Cyclohexa-1,3-diene (46 µl, 0.48 mmol), (Pr₄N)IO₄ (70 mg, 0.19 mmol), and **5b** (87 mg, 0.48 mmol) in CHCl₃ (1.5 ml). Purification of the yellowish solid by prep. TLC (AcOEt) led to **15g/16g** (83 mg, 67%). Colourless crystals. M.p. 117–118° ((i-Pr)₂O). [α]₂₀²⁰ = -4 (CHCl₃, c = 0.9) IR (KBr): 2970, 2940, 2820, 1650, 1410, 1365, 1260, 1200, 1165, 1110, 1045, 935, 905, 835, 760, 720, 700. ¹H-NMR (400 MHz, C₆D₆): *Tables 3* and 4. ¹³C-NMR (20.1 MHz, CDCl₃): *Table 5*. Anal. calc. for C₁₅H₁₇NO₃ (259.31): C 69.48, H 6.61, N 5.40; found: C 69.7, H 6.5, N 5.4.

15a from **17** and **6a**. A stirred soln. of phenyl carbamate **17** (203 mg, 0.88 mmol) and **6a** (100 μ l, 1 mmol, 1.13 equiv.) in pyridine (1.5 ml) was kept at 35–40° for 10 d. Pyridine was evaporated and the solid residue purified by FC (AcOEt/Et₂O 98:2): **15a** (108 mg, 52%). Colourless crystals. M.p. 76–77° (AcOEt/cyclohexane 2:8). [α]_D²⁰ = +3 (CHCl₃, c = 1.0). Spectral data: see above.

15b from (+)-**1** and **6b**. To a stirred soln. of (+)-**1** (187 mg, 1.26 mmol) in CH₂Cl₂ (4 ml) under Ar at r.t. were added Et₃N (211 µl, 1.51 mmol, 1.2 equiv.) and dropwise 0.45M phosgene in toluene (3.4 ml, 1.53 mmol, 1.2 equiv.). After 1 night at r.t., the soln. was evaporated, taken up in CCl₄, filtered, and evaporated and the oily residue dissolved in AcOEt (2 ml). To this stirred soln. were added **6b** (130 µl, 1.05 mmol, 1 equiv.) and K₂CO₃ (100 mg, 0.73 mmol). After 1 night at r.t., the mixture was filtered, the residue washed several times with hot AcOEt, and the soln. evaporated. The residue was purified by FC (AcOEt/cyclohexane 5:5): **15b** (158 mg, 60%). Colourless oil. $[\alpha]_{D}^{20} = -27.4$ (CHCl₃, c = 0.7). Spectral data: see above.

15d from (+)-1 and **8**. To a stirred soln. of **8** (50 µl, 0.34 mmol) in AcOEt (0.5 ml) at r.t. under Ar were added (+)-1 (58 mg, 0.39 mmol, 1.16 equiv.), K₂CO₃ (51 mg, 0.37 mmol, 1.1 equiv.), and H₂O (10 µl). After 5 h at r.t., the mixture was filtered, the residue washed several times with hot AcOEt, the org. soln. evaporated, and the residue purified by FC (AcOEt): **15d** (83 mg, 93%) which contains *ca*. 8% of **16d**. $[\alpha]_{D}^{20} = +18$ (CHCl₃, *c* = 0.7). IR (CCl₄): 2960, 2940, 2880, 1745, 1645, 1405, 1360, 1190, 1160, 920, 875. Spectral data: see above.

15e from (+)-**1** and **12**. To a stirred soln. of (+)-**1** (60 mg, 0.41 mmol) and Et₃N (53 µl, 0.41 mmol) in CH₂Cl₂ (2 ml) under Ar at r.t. were added **12** (115 mg, 0.53 mmol, 1.3 equiv.) and DCC (97 mg, 0.47 mmol, 1.15 equiv.). After 4 h at r.t., the mixture was filtered, the residue washed several times with CH₂Cl₂, the combined org. soln. evaporated, and the residue purified by prep. TLC (AcOEt): **15e** (115 mg, 92%). Colourless crystals. M.p. 112–114° ((i-Pr)₂O). [α]_D²⁰ = -65 (CHCl₃, c = 1.0). IR (KBr): 2960, 2930, 2870, 1685, 1650, 1610, 1390, 1360, 1160, 1120, 1070, 950, 905, 695. ¹H-NMR (250 MHz, CDCl₃, 318 K; 2 rotamers): 6.62 (*ddd*, H–C(5)); 6.50 (*ddd*, H–C(6)); 5.18 (*m*, H–C(4)); 4.71 (*m*, H–C(1)); 4.62, 4.50 (2 *m*, H–C(2'), 2 rotamers); 3.51, 3.39 (2 *m*, 2 H–C(5'), 2 rotamers); 2.21–1.65 (*m*, 2 H–C(4'), 2 H–C(7), 2 H–C(7), 2 H–C(8)); 1.46, 1.43 (2*s*, *t*-Bu, 2 rotamers). Anal. calc. for C₁₆H₂₄N₂O₄ (308.38); C 62.31, H 7.85, N 9.09; found: C 62.3, H 7.9, N 9.0.

15f from (+)-1 and **13a**. To a stirred soln. of **13a** (52 mg, 0.34 mmol) in abs. EtOH (0.8 ml) were added sequentially (+)-1 (50 mg, 0.34 mmol, 1 equiv.). Et₃N (48 µl, 0.35 mmol, 1 equiv.) and a soln. of DCC (74 mg, 0.36 mmol, 1.07 equiv.) in CHCl₃ (0.3 ml). After 1 night at r.t. under Ar, the mixture was evaporated, the residue taken up in Et₂O, the resulting suspension filtered, the residue washed with Et₂O and AcOEt, the combined org. soln. evaporated, and the residue purified by FC (AcOEt/cyclohexane 1:1): **15f** (57 mg, 68%). Colourless resin. $[\alpha]_{D}^{20} = -33$ (CHCl₃, c = 1.0). Spectral data: see above.

15g from (+)-1 and **13b**. To a stirred soln. of **13b** (98 mg, 0.59 mmol) in CH₂Cl₂(2 ml) at r.t. was added 1 drop of DMF and then dropwise oxalyl chloride (56 µl, 0.65 mmol, 1.1 equiv.; immediate reaction). After evaporation, the oily residue was dissolved in AcOEt (5 ml) at r.t. To this stirred soln. were added (+)-1 (97 mg, 0.66 mmol, 1.1 equiv.), K₂CO₃ (85 mg, 0.62 mmol, 1.04 equiv.), and H₂O (100 µl). After 4 h at r.t., the mixture was filtered, the residue washed several times with hot AcOEt, and the combined org. soln. evaporated. The residue was purified by FC (AcOEt): **15g** (100 mg, 65%). Colourless crystals. M.p. 103–104° ((i-Pr)₂O). [α]_D²⁰ = -48.8 (CHCl₃, *c* = 1.0). IR (KBr): 3050, 2935, 1640, 1410, 1365, 1255, 1185, 1095, 945, 745, 700. ¹H-NMR (250 MHz, C₆D₆, 343 K): 7.65 (*d*, 2 H_o); 7.13 (*m*, H_p, 2 H_m); 6.16 (*ddd*, H–C(5)); 5.94 (*ddd*, H–C(6)); 5.14 (*s*, H-C(2')); 5.05 (*m*, H–C(4)); 4.11 (*m*, H–C(1)); 3.30 (*s*, MeO); 1.49 (*m*, H_a–C(8), H_a–C(7)); 0.78 (*m*, H_b–C(8), H_b–C(7)). Anal. calc. for C₁₅H₁₇NO₃ (259,31): C 69.48, H 6.61, N 5.40; found: C 69.5, H 6.6, N 5.3.

(1 R,4 S)-3-/ (*Phenyloxy*) carbony/]-2-oxa-3-azabicyclo[2.2.2]oct-5-ene (17). To a stirred soln. of (+)-1 (433 mg, 2.9 mmol) in Et₂O (4.3 ml) at r.t. under Ar were added K₂CO₃ (436 mg, 3.15 mmol, 1.07 equiv.) and benzyl chloroformate (0.4 ml, 3.2 mmol, 1.1 equiv.). After 1 night at r.t., the mixture was filtered, the residue washed several times with hot AcOEt, the combined org. soln. evaporated, and the remaining semi-crystalline residue washed twice with Et₂O: 17 (512 mg, 75%). Colourless crystals. M.p. 130 ·132° (AcOEt). [α]₂₀²⁰ = +12.7 (CHCl₃, c = 1.0). IR (KBr): 1720, 1590, 1490, 1385, 1195, 1065, 1035, 995, 920, 880, 750, 690. ¹H-NMR (250 MHz, CDCl₃):

7.33 ($m, 2 H_m$); 7.20 (m, H_p); 7.10 ($m, 2 H_o$); 6.70, 6.64 (m, H-C(5), H-C(6)); 4.96 (m, H-C(4)); 4.85 (m, H-C(1)); 2.26 ($m, H_a-C(7), H_a-C(8)$); 1.55 ($m, H_b-C(7), H_b-C(8)$). Anal. calc. for C₁₃H₁₃NO₃ (231.25): C 67.52, H 5.67, N 6.06; found: C 67.2, H 5.7, N 6.1.

REFERENCES

- J. Hamer, M. Ahmad, in '1,4-Cycloaddition Reactions', Academic Press, New York, 1967; G. Kresze, J. Firl, Fortschr. Chem. Forsch. 1969, 11, 245; W. Seidenfaden, in 'Methoden der Organischen Chemie, Houben-Weyl', Thieme, Stuttgart, 1970, X-1, p. 1019; G. W. Kirby, Chem. Soc. Rev. 1977, 6, 1; T. L. Gilchrist, *ibid.* 1983, 12, 53; H. U. Reissig, Nachr. Chem. Tech. Lab. 1986, 34, 237.
- [2] N. V. Leonard, A. J. Playtis, F. Skoog, R. Y. Schmitz, J. Am. Chem. Soc. 1971, 93, 3056; B. Belleau, Yum-Kin Au-Young, *ibid.* 1963, 85, 64.
- [3] M. Jung, G. Offenbächer, J. Rétey, *Helv. Chim. Acta* 1983, 66, 1915; J. E. Baldwin, P. D. Bailey, G. Gallacher, K. A. Singleton, P. M. Wallace, *J. Chem. Soc., Chem. Commun.* 1983, 1049; T. P. Burkholder, P. L. Fuchs, *J. Am. Chem. Soc.* 1988, 110, 2341.
- [4] a) G. Augelmann, J. Streith, H. Fritz, *Helv. Chim. Acta* 1985, 68, 95; b) A. Defoin, H. Fritz, Ch. Schmidlin, J. Streith, *ibid.* 1987, 70, 554; c) A. Defoin, H. Fritz, G. Geffroy, J. Streith, *ibid.* 1988, 71, 1642.
- [5] G. Kresze, E. Kysela, *Liebigs Ann. Chem.* 1981, 202; G. Kresze, H. Melzer, *ibid.* 1981, 1874; H. Braun, K. Klier, G. Kresze, M. Sabuni, O. Werbitzky, J. Winkler, *ibid.* 1986, 1360; G. Kresze, M. M. Weiss, W. Dittel, *ibid.* 1984, 203; G. Kresze, W. Dittel, *ibid.* 1981, 610; G. Kresze, W. Dittel, H. Melzer, *ibid.* 1981, 224; G. Kresze, E. Kysela, W. Dittel, *ibid.* 1981, 210.
- [6] O. Werbitzky, K. Klier, H. Felber, Liebigs Ann. Chem. 1990, 267.
- [7] M. Sabuni, G. Kresze, H. Braun, Tetrahedron Lett. 1984, 25, 5377.
- [8] H. Felber, G. Kresze, R. Prewo, A. Vasella, *Helv. Chim. Acta* 1986, 69, 1137; H. Felber, G. Kresze, H. Braun, A. Vasella, *Tetrahedron Lett.* 1984, 25, 5381.
- [9] H. Braun, R. Charles, G. Kreszc, M. Sabuni, J. Winkler, Liebigs Ann. Chem. 1987, 1129.
- [10] G. E. Keck, R. R. Webb, J. B. Yates, *Tetrahedron* 1981, 37, 4007; G. W. Kirby, J. G. Sweeny, J. Chem. Soc., Chem. Commun. 1973, 704.
- [11] V. Gouverneur, L. Ghosez, Tetrahedron: Asymmetry 1990, 1, 363.
- [12] A. Defoin, A. Brouillard-Poichet, J. Streith, Helv. Chim. Acta 1991, 74, 103.
- [13] A. Brouillard-Poichet, A. Defoin, J. Streith, Tetrahedron Lett. 1989, 30, 7061.
- [14] G. W. Kirby, M. Nazeer, Tetrahedron Lett. 1988, 29, 6173.
- [15] A. Miller, T. Mc C. Paterson, G. Procter, Synlett 1989, 32.
- [16] G. Zinner, G. Isensee, Arch. Pharm. (Weinheim) 1974, 307, 7.
- [17] G. M. Steinberg, J. Bolger, J. Org. Chem. 1956, 21, 660; P. Gröbner, E. Müller, Eur. J. Med. Chem. Theor. 1974, 9, 341.
- [18] C. R. Hauser, W. B. Renfrow, Jr., Org. Synth. Coll. Vol. II 1963, 67.
- [19] St. F. Nelsen, J. A. Thompson-Colon, B. Kirste, A. Rosenhouse, M. Kaftory, J. Am. Chem. Soc. 1987, 109, 7128.
- [20] D. Geffken, H.J. Kämpf, Synthesis 1975, 176.
- [21] H. Waldmann, M. Dräger, Liebigs Ann. Chem. 1990, 681; Tetrahedron Lett. 1989, 30, 4227.
- [22] W. Choy, L. A. Reed III, S. Masamune, J. Org. Chem. 1983, 48, 1137.
- [23] A. Miller, G. Proeter, Tetrahedron Lett. 1990, 31, 1041.
- [24] D. Enders, P. Fey, H. Kipphardt, Org. Synth. 1987, 65, 173.